



University of Saskatchewan

Department of Chemical Engineering  
ChE 311- Mathematical Modelling I

Quiz #3

DATE: Wednesday December 1, 2004  
INSTRUCTOR: Professor T. Pugsley  
TIME: 10:30 - 11:20 a.m., RM 1C70 Eng.

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**Instructions:** This is a closed book/closed notes quiz. Personal calculators are permitted. Write your answers neatly in the examination booklets provided. Please do both questions.

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**Question #1 (10 marks)**

A plant at Canso, Nova Scotia, makes fish-protein concentrate (FPC). One of the operating problems is the drying of the FPC. It dries in the fluidized dryer at a rate proportional to its moisture content. If a given batch of FPC loses one-half of its initial moisture in the first 15 min, how long will it take to remove 90% of the water in the batch of FPC?

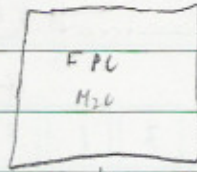
**Question #2 (10 marks)**

A sewage disposal plant has a big concrete holding tank of 100,000 gal capacity. It is  $\frac{3}{4}$ -full of liquid to start with and contains 60,000 lb of organic material in suspension. Water runs into the holding tank at a rate of 20,000 gal/h and the solution leaves at the rate of 15,000 gal/h. How much organic material is in the tank at the end of three hours?

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**END OF EXAMINATION**

① Given



→ 50%  $H_2O$  is lost after 15 min

Find → how long it will take to lose 90%  $H_2O$

Solution

→  $H_2O$  balance

$$Input = 0$$

$$Output = kX \text{ at}$$

$$Acc = (X)_{t_{\text{test}}} - (X)_t$$

$$Acc = Input - Output$$

$$\frac{dX}{dt} = 0 - kX$$

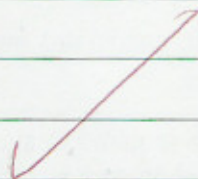
$$\frac{dX}{dt} = -kX$$

$$\int_{X_0}^X \frac{dX}{X} = -k \int_0^t dt$$

$$\ln \frac{X}{X_0} = -kt$$

$$\frac{X}{X_0} = e^{-kt}$$

$$X = X_0 e^{-kt}$$





→ known conditions

$$\rightarrow t = 15, \quad x = \frac{1}{2} x_0$$

$$x = x_0 e^{-kt}$$

$$\frac{1}{2} x_0 = x_0 e^{-15k}$$

$$\frac{1}{2} = e^{-15k}$$

$$-0.693 = -15k$$

$$k = 0.0462 \text{ min}^{-1}$$

→ time for  $x = 0.1 x_0$

$$x = x_0 e^{-kt}$$

$$0.1 x_0 = x_0 e^{-0.0462t}$$

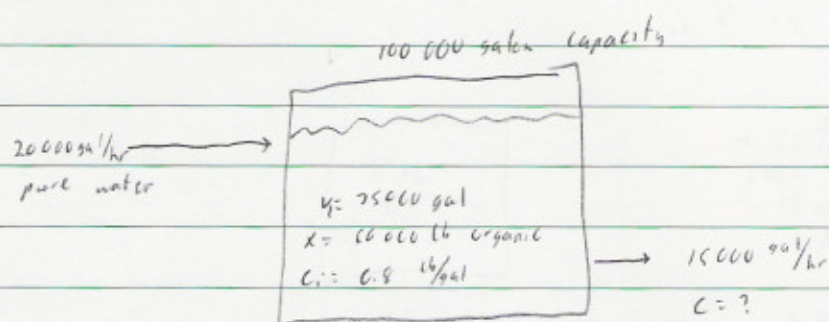
$$0.1 = e^{-0.0462t}$$

$$-2.30 = -0.0462t$$

$$t = 49.78$$

$$t = \boxed{49.8 \text{ min}}$$

② given



Find → how much organic material is in the tank at the end of 3 hours

Solution

→  $H_2O$  balance

$$\text{Input} = 20 000 \text{ At}$$

$$\text{Output} = 15 000 \text{ At}$$

$$\text{Acc} = (V)_{\text{final}} - (V)_i$$

$$\text{Acc} = \text{Input} - \text{Output}$$

$$\frac{dV}{dt} = 20 000 - 15 000$$

$$\frac{dV}{dt} = 5 000$$

$$\int_{V_i}^V dV = 5 000 \int_0^t dt$$

$$V - V_i = 5 000 t$$

$$V = 5 000 t + 75 000$$

→ at time  $t = 3 \text{ hr}$

$$V = 5 000 (3) + 75 000$$

$$V = 90 000 \text{ gal}$$



→ organic balance

$$I_{\text{input}} = 0$$

$$\text{output} = 15000 \text{ C at}$$

$$Acc = (VC)_{\text{final}} - (VC)_i$$

$$Acc = I_{\text{input}} - \text{output}$$

$$\frac{d(VC)}{dt} = 0 - 15000 \text{ C}$$

$$C \frac{dV}{dt} + V \frac{dC}{dt} = -15000 \text{ C}$$

$$\frac{dV}{dt} + \frac{V}{C} \frac{dC}{dt} = -15000$$

$$5000 + \frac{V}{C} \frac{dC}{dt} = -15000$$

$$\frac{V}{C} \frac{dC}{dt} = -20000$$

$$V \int_{C_i}^C \frac{dC}{C} = -20000 \int_0^t dt$$

$$V \ln \left( \frac{C}{C_i} \right) = -20000 t$$

but this is  
not constant

$V = 5000t + \text{const}$

→ at time  $t = 3 \text{ hr}$

$$75000 \ln \left( \frac{C}{0.8} \right) = -20000(3)$$

$$\ln \left( \frac{C}{0.8} \right) = -0.8$$

$$\left( \frac{C}{0.8} \right) = e^{-0.8}$$

$$C = 0.8 e^{-0.8}$$

$$C = 0.8(0.449)$$

$$C = 0.359 \text{ lb/gal}$$

$$x = CV$$

$$x = (0.359 \text{ lb/gal})(90000 \text{ gal})$$

$$x = 32310 \text{ lb}$$

$$x = \boxed{32300 \text{ lb}}$$